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WHAT IS CLAIMED IS:

- 1. A method of improving etch selectivity of silicon nitride relative to an adjacent patterned organic DUV photoresist during plasma etching, comprising: reacting a surface of said patterned organic DUV photoresist with plasma species generated from a plasma source gas consisting essentially of at least one inorganic fluorine-comprising gas and sulfur dioxide (SO₂), wherein the molecular ratio of said inorganic fluorine-comprising gas to said sulfur dioxide ranges from about 50: 1 to about 1: 1, to produce a reacted DUV photoresist surface, whereby the etch rate of said organic DUV photoresist is slowed, while said silicon nitride exposed through said patterned organic DUV photoresist is etched.
 - 2. The method of Claim 1, wherein the amount of sulfur dioxide (SO₂) present in said plasma source gas is varied during said etching of said silicon nitride.
 - 3. The method of Claim 1, wherein one of said at least one inorganic fluorine-comprising gases is sulfur hexafluoride (SF₆).
- 1 4. The method of Claim 1, wherein one of said at least one inorganic fluorine-2 comprising gases is nitrogen trifluoride (NF₃).
- The method of Claim 1, wherein said plasma source gas includes two inorganic fluorine-comprising gases.
- 1 6. The method of Claim 1, wherein a temperature of a substrate which includes said silicon nitride is between about 20°C and 100°C during exposure to said plasma etchant.

- The method of Claim 6, wherein said temperature is between about 40°C and
- 2 60°C.
- 1 8. The method of Claim 1, wherein said silicon nitride etch rate is at least two times
- 2 said organic DUV photoresist etch rate.
- 9. The method of Claim 1, wherein said patterned organic DUV photoresist has a thickness of less than about 4000 Å.
- 1 10. The method of Claim 1, wherein said exposing of the structure comprising said 2 silicon nitride and said patterned organic DUV photoresist to said plasma etchant results in 3 an etched inorganic substrate having a feature size less than 2500 Å.
- 1 11. The method of Claim 1, wherein said selectivity of said silicon nitride relative to said adjacent patterned organic DUV photoresist is greater than 1.5.
- 1 12. The method of Claim 11, wherein said selectivity is greater than about 2.0.
- 1 13. The method of Claim 1, wherein said plasma etchant is generated from a high density plasma.
- 1 14. The method of Claim 13, wherein said plasma density is at least 10^{11} e^{-/cm³}.
- 1 15. A method of improving etch selectivity of silicon nitride relative to an adjacent 2 patterned organic DUV photoresist during plasma etching, comprising: reacting a surface 3 of said patterned organic DUV photoresist with plasma species generated from a plasma

- 4 source gas consisting essentially of at least one inorganic fluorine-comprising gas, sulfur
- 5 dioxide (SO₂), and a diluent gas selected from the group consisting of Ar, Kr, Xe, and He,
- 6 wherein the molecular ratio of said inorganic fluorine-comprising gas to said sulfur dioxide
- 7 ranges from about 50: 1-to about 1:1, to produce a reacted DUV photoresist surface,
- 8 whereby the etch rate of said organic DUV photoresist is slowed, while said silicon nitride
- 9 exposed through said patterned organic DUV photoresist is etched.
- 1 16. The method of Claim 15, wherein the amount of sulfur dioxide (SO₂) present in said plasma source gas is varied during said etching of said silicon nitride.
- 1 17. The method of Claim 15, wherein a molecular ratio of said inorganic fluorinecomprising gas to said diluent gas ranges from about 0.1:1 to about 10:1.
- 1 18. The method of Claim 15, wherein said diluent gas is argon (Ar).
- 1 19. The method of Claim 15, wherein said sulfur dioxide amount is about 2% 20% 2 by volume of said plasma source gas.
- 1 20. The method of Claim 19, wherein said sulfur dioxide amount is about 10% -15%
- 2 by volume of said plasma source gas.
- 1 21. The method of Claim 15, wherein said at least one inorganic fluorine-comprising
- 2 gas amount is about 20% 60% by volume of said plasma source gas.
- The method of Claim 21, wherein said at least one inorganic fluorine-comprising
- 2 gas amount is about 25% 35% by volume of said plasma source gas.

- 1 23. The method of Claim 18, wherein said argon amount is about 20% 60% by
- 2 volume of said plasma source gas.
- 1 24. The method of Claim 23, wherein said argon amount is about 50% 60% by
- 2 volume of said plasma source gas.
- The method of Claim 15, wherein one of said at least one inorganic fluorinecomprising gases is sulfur hexafluoride (SF₆).
 - 26. The method of Claim 15, wherein one of said at least one inorganic fluorine-comprising gases is nitrogen trifluoride (NF₃).
 - 27. The method of Claim 15, wherein said plasma source gas includes two inorganic fluorine-comprising gases.
- 1 28. The method of Claim 27, wherein a total amount of said inorganic fluorine-2 comprising gases is about 20% - 60% by volume of the plasma source gas.
- 29. A method of improving etch selectivity of silicon nitride relative to an adjacent patterned organic DUV photoresist during plasma etching, comprising: reacting a surface of said patterned organic DUV photoresist with plasma species generated from a plasma source gas consisting essentially of at least one inorganic fluorine-comprising gas, sulfur dioxide (SO₂), and hydrogen bromide (HBr), wherein the molecular ratio of said inorganic fluorine-comprising gas to said sulfur dioxide ranges from about 50 : 1 to about 1 : 1, to produce a reacted DUV photoresist surface, whereby the etch rate of said organic DUV

- 8 photoresist is slowed, while said silicon nitride exposed through said patterned organic DUV
- 9 photoresist is etched.
- 1 30. The method of Claim 29, wherein the amount of sulfur dioxide (SO₂) present in
- 2 said plasma source gas is varied during said etching of said silicon nitride.
- 1 31. The method of Claim 29, wherein said sulfur dioxide amount is about 2% 20%
- 2 by volume of said plasma source gas.
- The method of Claim 31, wherein said sulfur dioxide amount is about 10% 15%
- by volume of said plasma source gas.
 - The method of Claim 29, wherein said at least one inorganic fluorine-comprising
- gas amount is about 20% 60% by volume of said plasma source gas.
- 1 34. The method of Claim 33, wherein said at least one inorganic fluorine-comprising
- 2 gas amount is about 25% 35% by volume of said plasma source gas.
- The method of Claim 29, wherein said hydrogen bromide amount is about
- 2 10% 60% by volume of said plasma source gas.
- 1 36. The method of Claim 35, wherein said hydrogen bromide amount is about
- 2 20% 40% by volume of said plasma source gas.
- The method of Claim 29, wherein one of said at least one inorganic fluorine-
- 2 comprising gases is sulfur hexafluoride (SF₆).

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- The method of Claim 29, wherein one of said at least one inorganic fluorinecomprising gases is nitrogen trifluoride (NF₃).
- 1 39. The method of Claim 29, wherein said plasma source gas includes two inorganic fluorine-comprising gases.
- 1 40. The method of Claim 39, wherein a total amount of said inorganic fluorine-2 comprising gases is about 20% - 60% by volume of the plasma source gas.
 - A method of improving etch selectivity of silicon nitride relative to an adjacent patterned organic DUV photoresist during plasma etching, comprising: reacting a surface of said patterned organic DUV photoresist with plasma species generated from a plasma source gas consisting essentially of at least one inorganic fluorine-comprising gas, sulfur dioxide (SO₂), hydrogen bromide (HBr), and a diluent gas selected from the group consisting of Ar, Kr, Xe, and He, wherein the molecular ratio of said inorganic fluorine-comprising gas to said sulfur dioxide ranges from about 50: 1 to about 1:1, to produce a reacted DUV photoresist surface, whereby the etch rate of said organic DUV photoresist is slowed, while said silicon nitride exposed through said patterned organic DUV photoresist is etched.
- 1 42. The method of Claim 41, wherein the amount of sulfur dioxide (SO₂) present in 2 said plasma source gas is varied during said etching of said silicon nitride.
- 1 43. The method of Claim 41, wherein a molecular ratio of said inorganic fluorinecomprising gas to said diluent gas ranges from about 0.1:1 to about 10:1.

- 1 44. The method of Claim 41, wherein said diluent gas is argon (Ar).
- 1 45. The method of Claim 41, wherein said sulfur dioxide amount is about 2% 20%
- 2 by volume of said plasma source gas.
- 1 46. The method of Claim 45, wherein said sulfur dioxide amount is about 10% -15%
- 2 by volume of said plasma source gas.
- 1 47. The method of Claim 41, wherein said at least one inorganic fluorine-comprising
- 2 gas amount is about 20% 60% by volume of said plasma source gas.
- 1 48. The method of Claim 47, wherein said at least one inorganic fluorine-comprising
- 2 gas amount is about 25% 35% by volume of said plasma source gas.
- 1 49. The method of Claim 41, wherein said hydrogen bromide amount is about
- 2 10% 60% by volume of said plasma source gas.
- 1 50. The method of Claim 49, wherein said hydrogen bromide amount is about
- 2 20% 40% by volume of said plasma source gas.
- 1 51. The method of Claim 44, wherein said argon amount is about 20% 60% by
- 2 volume of said plasma source gas.
- 1 52. The method of Claim 51, wherein said argon amount is about 50% 60% by
- 2 volume of said plasma source gas.

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- The method of Claim 41, wherein one of said at least one inorganic fluorine-53. 1 comprising gases is sulfur hexafluoride (SF₆). 2
- The method of Claim 41, wherein one of said at least one inorganic fluorine-54. 1 comprising gases is nitrogen trifluoride (NF₃). 2
- The method of Claim 41, wherein said plasma source gas includes two inorganic 55. fluorine-comprising gases.
 - The method of Claim 55, wherein a total amount of said inorganic fluorine-56. comprising gases is about 20% - 60% by volume of the plasma source gas.
 - A method of improving etch selectivity of silicon nitride relative to an adjacent 57. patterned organic DUV photoresist during plasma etching, comprising: reacting a surface of said patterned organic DUV photoresist with plasma species generated from a plasma source gas consisting essentially of sulfur hexafluoride (SF₆) and sulfur dioxide (SO₂), wherein the molecular ratio of said sulfur hexafluoride to said sulfur dioxide ranges from about 50:1 to about 1:1, to produce a reacted DUV photoresist surface, whereby the etch rate of said organic DUV photoresist is slowed, while said silicon nitride exposed through said patterned organic DUV photoresist is etched.
 - The method of Claim 57, wherein the amount of sulfur dioxide (SO₂) present in 1 58. said plasma source gas is varied during said etching of said silicon nitride. 2

- A method of improving etch selectivity of silicon nitride relative to an adjacent 1 59. patterned organic DUV photoresist during plasma etching, comprising: reacting a surface 2 of said patterned organic DUV photoresist with plasma species generated from a plasma 3 source gas consisting essentially of sulfur hexafluoride (SF₆), sulfur dioxide (SO₂), and 4 argon (Ar), wherein the molecular ratio of said sulfur hexafluoride to said sulfur dioxide 5 ranges from about 50:1 to about 1:1, to produce a reacted DUV photoresist surface, 6 whereby the etch rate of said organic DUV photoresist is slowed, while said silicon nitride 7 exposed through said patterned organic DUV photoresist is etched. 8
- 60. The method of Claim 59, wherein the amount of sulfur dioxide (SO₂) present in said plasma source gas is varied during said etching of said silicon nitride.
 - 61. The method of Claim 59, wherein a molecular ratio of said sulfur hexafluoride to said argon ranges from about 0.1 : 1 to about 10 : 1.
 - 62. The method of Claim 59, wherein said sulfur dioxide amount is about 2% 20% by volume of said plasma source gas.
- 1 63. The method of Claim 62, wherein said sulfur dioxide amount is about 10% -15% by volume of said plasma source gas.
- 1 64. The method of Claim 59, wherein said sulfur hexafluoride amount is about 20% 60% by volume of said plasma source gas.
- 1 65. The method of Claim 64, wherein said sulfur hexafluoride amount is about 25% 2 35% by volume of said plasma source gas.

- 1 66. The method of Claim 59, wherein said argon amount is about 20% 60% by volume of said plasma source gas.
- 1 67. The method of Claim 66, wherein said argon amount is about 50% 60% by volume of said plasma source gas.
 - A method of improving etch selectivity of silicon nitride relative to an adjacent patterned organic DUV photoresist during plasma etching, comprising: reacting a surface of said patterned organic DUV photoresist with plasma species generated from a plasma source gas consisting essentially of sulfur hexafluoride (SF₆), sulfur dioxide (SO₂), and hydrogen bromide (HBr), wherein the molecular ratio of said inorganic fluorine-comprising gas to said sulfur dioxide ranges from about 50:1 to about 1:1, to produce a reacted DUV photoresist surface, whereby the etch rate of said organic DUV photoresist is slowed, while said silicon nitride exposed through said patterned organic DUV photoresist is etched.
- 1 69. The method of Claim 68, wherein the amount of sulfur dioxide (SO₂) present in 2 said plasma source gas is varied during said etching of said silicon nitride.
- The method of Claim 68, wherein said sulfur dioxide amount is about 2% 20% by volume of said plasma source gas.
- The method of Claim 70, wherein said sulfur dioxide amount is about 10% 15% by volume of said plasma source gas.

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- 1 72. The method of Claim 68, wherein said sulfur hexafluoride amount is about
- 2 20% 60% by volume of said plasma source gas.
- The method of Claim 72, wherein said sulfur hexafluoride amount is about
- 2 25% 35% by volume of said plasma source gas.
- The method of Claim 68, wherein said hydrogen bromide amount is about
- 2 10% 60% by volume of said plasma source gas.
- 1 75. The method of Claim 74, wherein said hydrogen bromide amount is about
- 2 20% 40% by volume of said plasma source gas.
 - A method of improving etch selectivity of silicon nitride relative to an adjacent patterned organic DUV photoresist during plasma etching, comprising: reacting a surface of said patterned organic DUV photoresist with plasma species generated from a plasma source gas consisting essentially of sulfur hexafluoride (SF₆), sulfur dioxide (SO₂), hydrogen bromide (HBr), and a diluent gas selected from the group consisting of Ar, Kr, Xe, and He, wherein the molecular ratio of said inorganic fluorine-comprising gas to said sulfur dioxide ranges from about 50: 1 to about 1:1, to produce a reacted DUV photoresist surface,

whereby the etch rate of said organic DUV photoresist is slowed, while said silicon nitride

- 9 exposed through said patterned organic DUV photoresist is etched.
- The method of Claim 76, wherein the amount of sulfur dioxide (SO₂) present in
- 2 said plasma source gas is varied during said etching of said silicon nitride.

- 1 78. The method of Claim 76, wherein a molecular ratio of said inorganic fluorine-
- 2 comprising gas to said diluent gas ranges from about 0.1:1 to about 10:1.
- 79. 1 The method of Claim 76, wherein said diluent gas is argon (Ar).
- 80. The method of Claim 76, wherein said sulfur dioxide amount is about 2% - 20% 1
- 2 by volume of said plasma source gas.
- The method of Claim 80, wherein said sulfur dioxide amount is about 10% -15% 81. 1 2 2 in the second of the sec
 - by volume of said plasma source gas.
 - The method of Claim 76, wherein said sulfur hexafluoride amount is about 82.
 - 20% 60% by volume of said plasma source gas.
 - The method of Claim 82, wherein said sulfur hexafluoride amount is about 83.
 - 25% 35% by volume of said plasma source gas.
 - The method of Claim 76, wherein said hydrogen bromide amount is about 1 84.
 - 10% 60% by volume of said plasma source gas. 2
 - The method of Claim 84, wherein said hydrogen bromide amount is about 1 85.
 - 20% 40% by volume of said plasma source gas. 2
 - The method of Claim 79, wherein said argon amount is about 20% 60% by 1 86.
 - 2 volume of said plasma source gas.

- 1 87. The method of Claim 86, wherein said argon amount is about 50% 60% by
- 2 volume of said plasma source gas.